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Advanced Oxidation Processes as Alternative to Standard Whitening Methods

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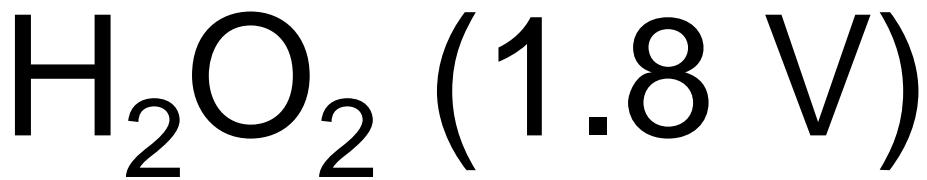
**THE REMOVAL OF STAINS FROM THE TEETH
CAUSED BY THE ADMINISTRATION OF MED-
ICINAL AGENTS AND THE BLEACHING OF
PULPLESS TEETH.**

BY A. W. HARLAN, M.D., CHICAGO, ILL.

Read in the Section of Oral and Dental Surgery of the American Medical Association, May, 1884.

GENTLEMEN:—A large number of remedial agents administered by physicians temporarily stain the teeth, but in looking over the list I find there are but few which may be said to permanently stain them. The mineral acids—nitric, sulphuric, hydrochloric, and other acids of this nature, if used for any length of time, may discolor the teeth and likewise have a deleterious effect on them; yet it cannot be said that such agents stain the teeth so that any particular method should be desired for restoring their natural appearance. The vegetable series may likewise

JAMA, 1885; IV(5):123-125



(Oppenländer T. 2003)



Effectiveness of Light Sources on In-Office Dental Bleaching: A Systematic Review and Meta- Analyses

JR SoutoMaior • SLD de Moraes • CAA Lemos
BC do E Vasconcelos • MAJR Montes • EP Pellizzer

Operative Dentistry. May/June 2019, Vol. 44, No. 3.

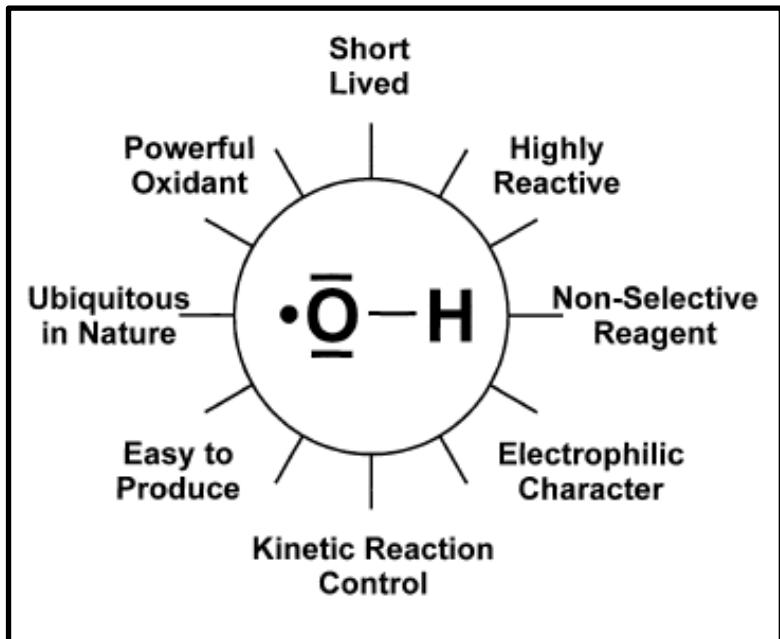


Introduction

AOPs: Advanced Oxidation Processes



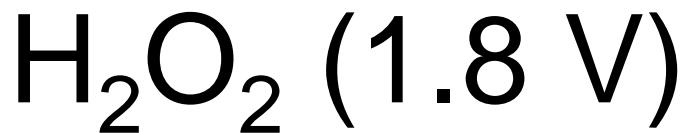
AOPs: Advanced Oxidation Processes



(Oppenländer T. 2003)

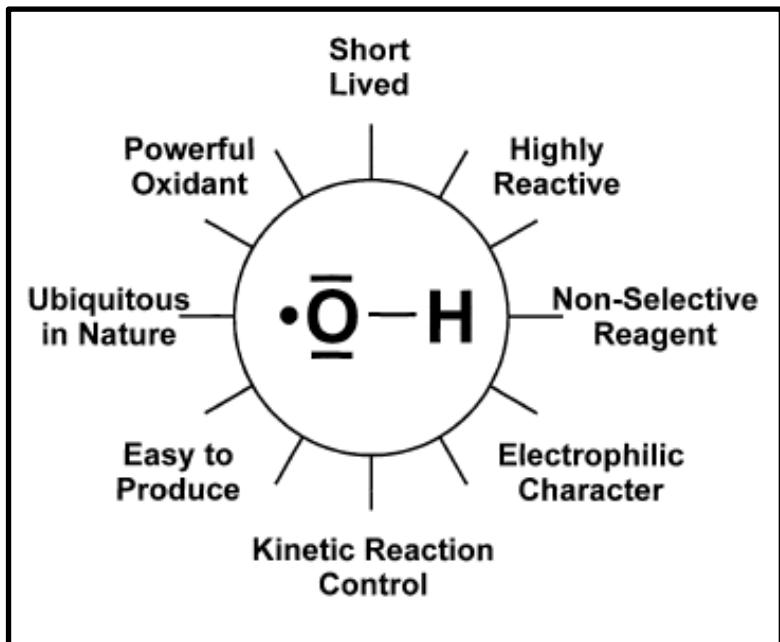


X



Whitening Method in Dentistry

AOPs: Advanced Oxidation Processes



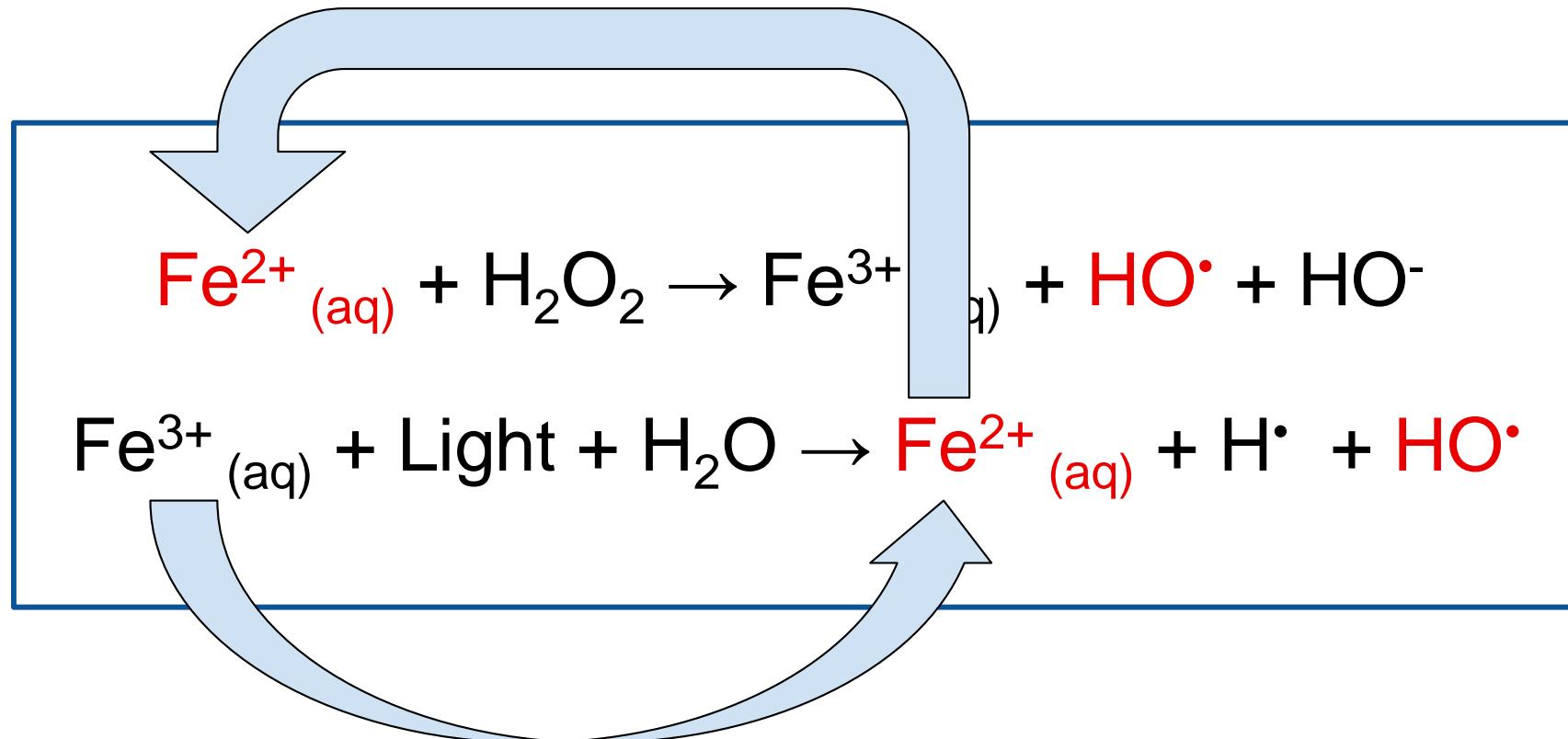
(Oppenländer T. 2003)

Hydroxyl ($\text{OH}\cdot$) based AOPs

- $\text{H}_2\text{O}_2 + \text{UV}$
- $\text{O}_3 + \text{UV}$
- $\text{H}_2\text{O}_2 + \text{Fe}$
- $\text{H}_2\text{O}_2 + \text{Fe} + \text{Light}$
- $\text{HO}_2 + \text{UV}$

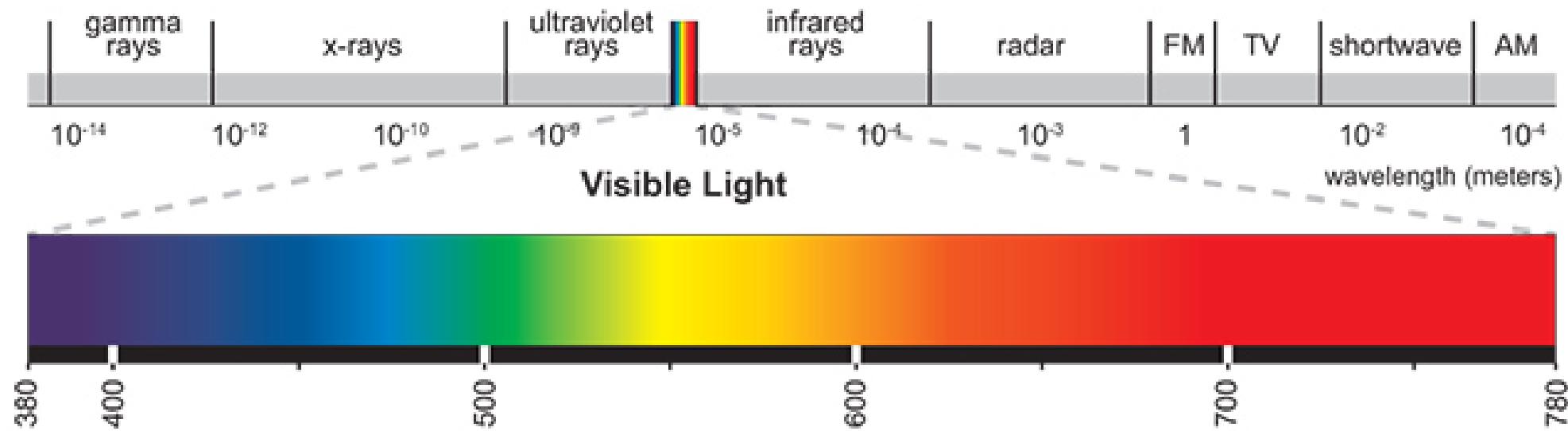
AOPs: Advanced Oxidation Processes

Photo-Fenton

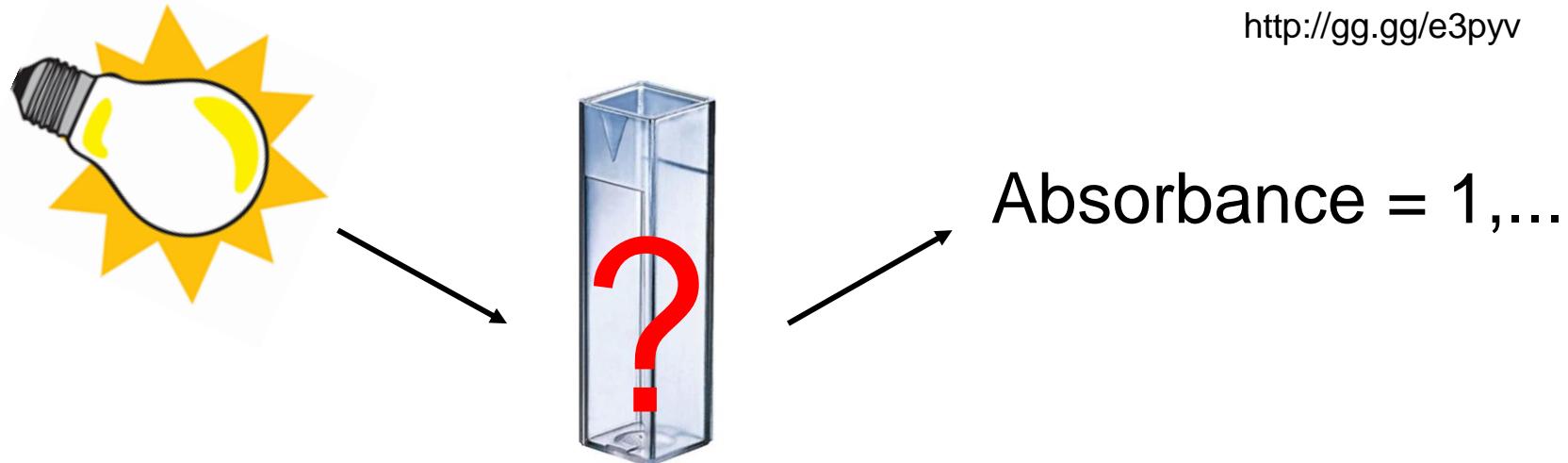


Verify if AOPs (Fenton and photo-Fenton) are more efficient than conventional whitening methods (HP or HP + light).

Spectrophotometry



<http://gg.gg/e3pyv>



<http://dx.doi.org/10.1590/1678-775720130578>

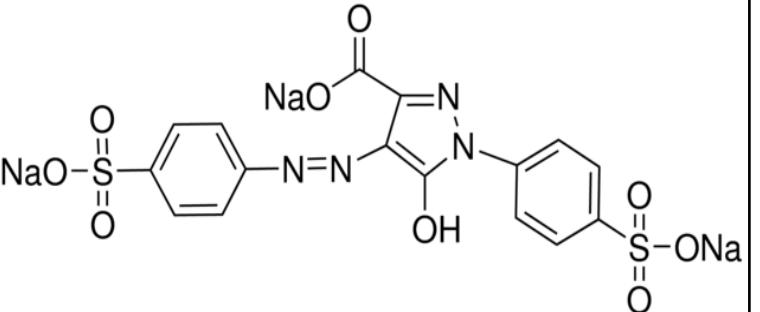
Effect of coffee and a cola-based soft drink on the color stability of bleached bovine incisors considering the time elapsed after bleaching

Rodrigo PIROLO¹, Rafael Francisco Lia MONDELLI², Gisele Maria CORRER³, Carla Castiglia GONZAGA³, Adilson Yoshio FURUSE²

Journal of Applied Oral Science. May 2014.

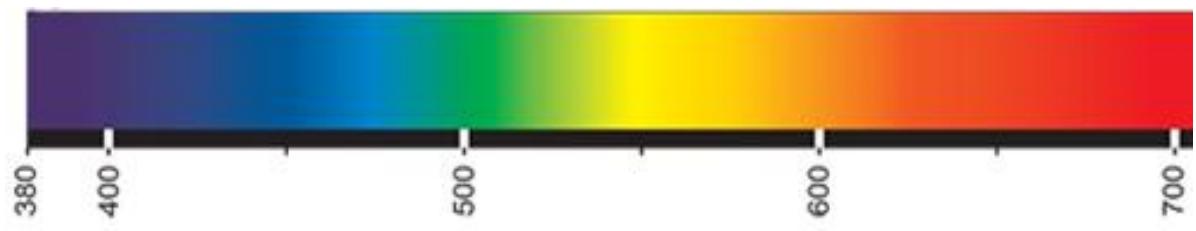
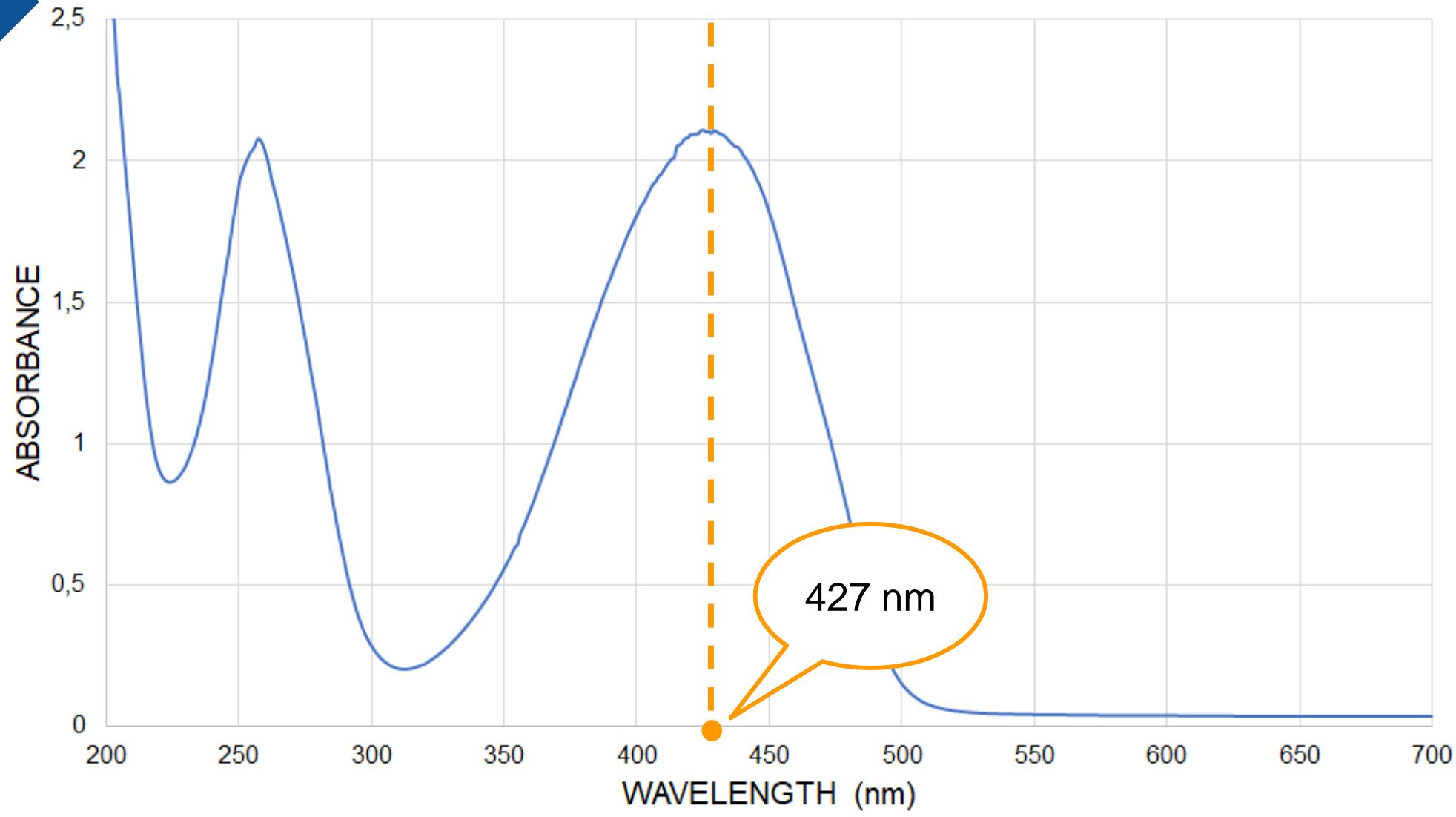




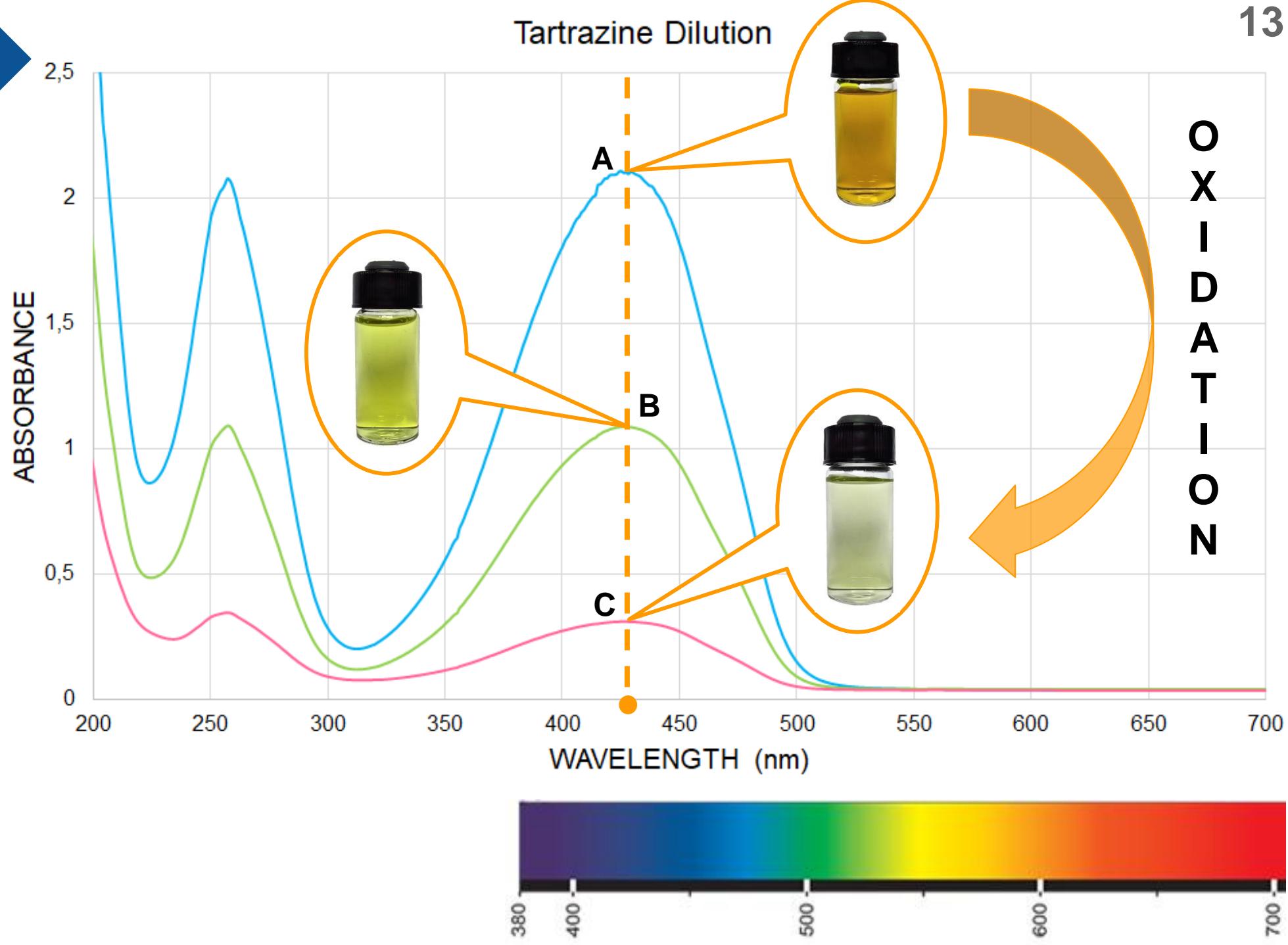
Name	Structural formula	Molecular weight (g/mol)	Absorbance (nm)
Tartrazine		534,3	427

Methods

Tartrazine Dilution



Methods

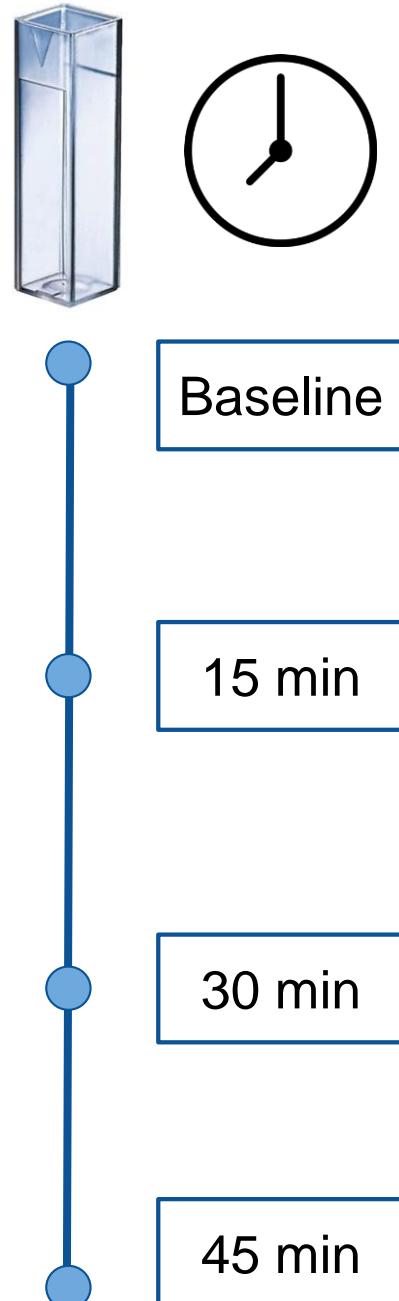


Methods

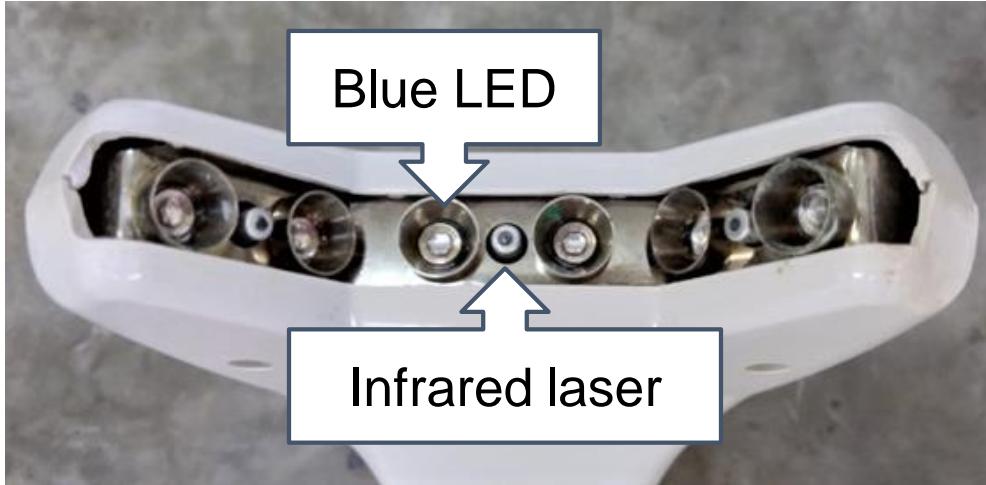
OXIDANT	OXIDANT CONCENTRATION	LIGHT PRESENCE		IRON PRESENCE	
HP	40%				
	25%	Present	Absent	Present	Absent
	5%	✓	✗	✓	✗

Experimental Groups

	Group	HP	Light	Iron
HP	HP 40%	[40]	X	X
	HP 25%	[25]	X	X
	HP 5%	[5]	X	X
HP + Light	HP 40% + Light	[40]	✓	X
	HP 25% + Light	[25]	✓	X
	HP 5% + Light	[5]	✓	X
HP + Iron	HP 40% + Fe	[40]	X	✓
	HP 25% + Fe	[25]	X	✓
	HP 5% + Fe	[5]	X	✓
HP + Light + Iron	HP 40% + Light + Fe	[40]	✓	✓
	HP 25% + Light + Fe	[25]	✓	✓
	HP 5% + Light + Fe	[5]	✓	✓



Methods

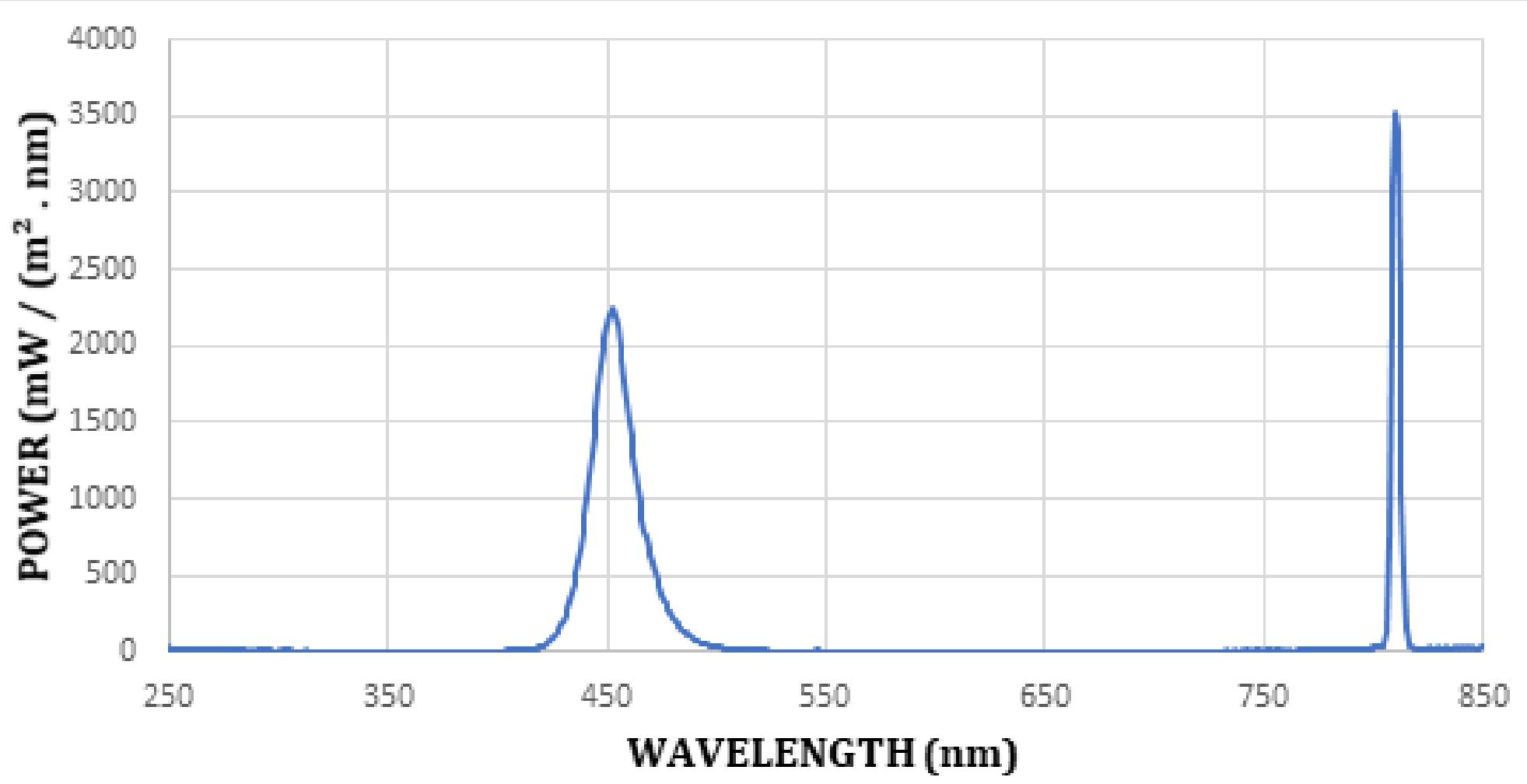


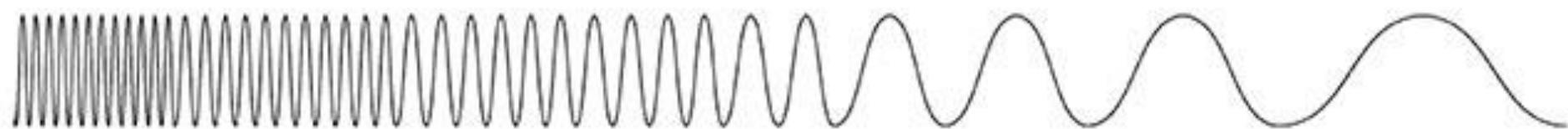
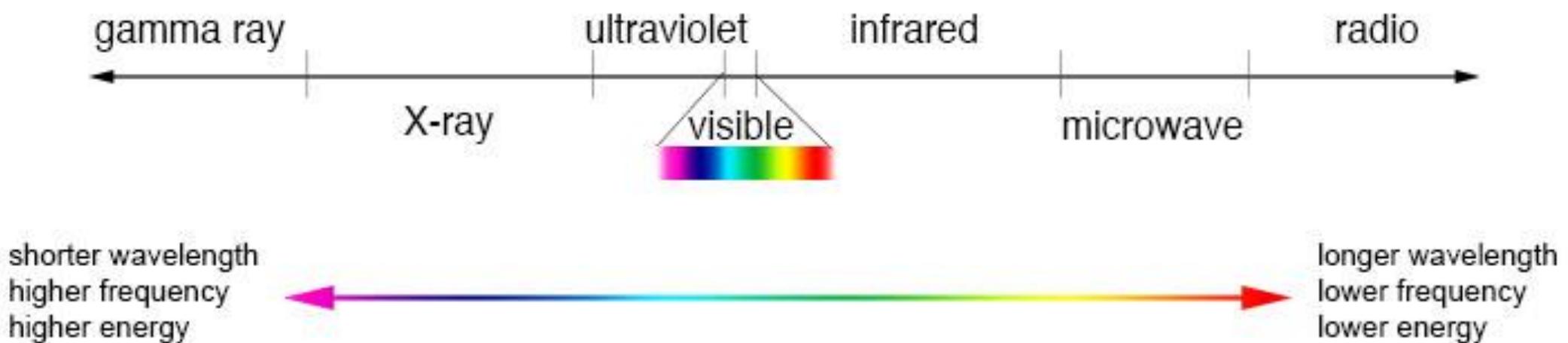
Results



97.3 °F (36.3 °C)

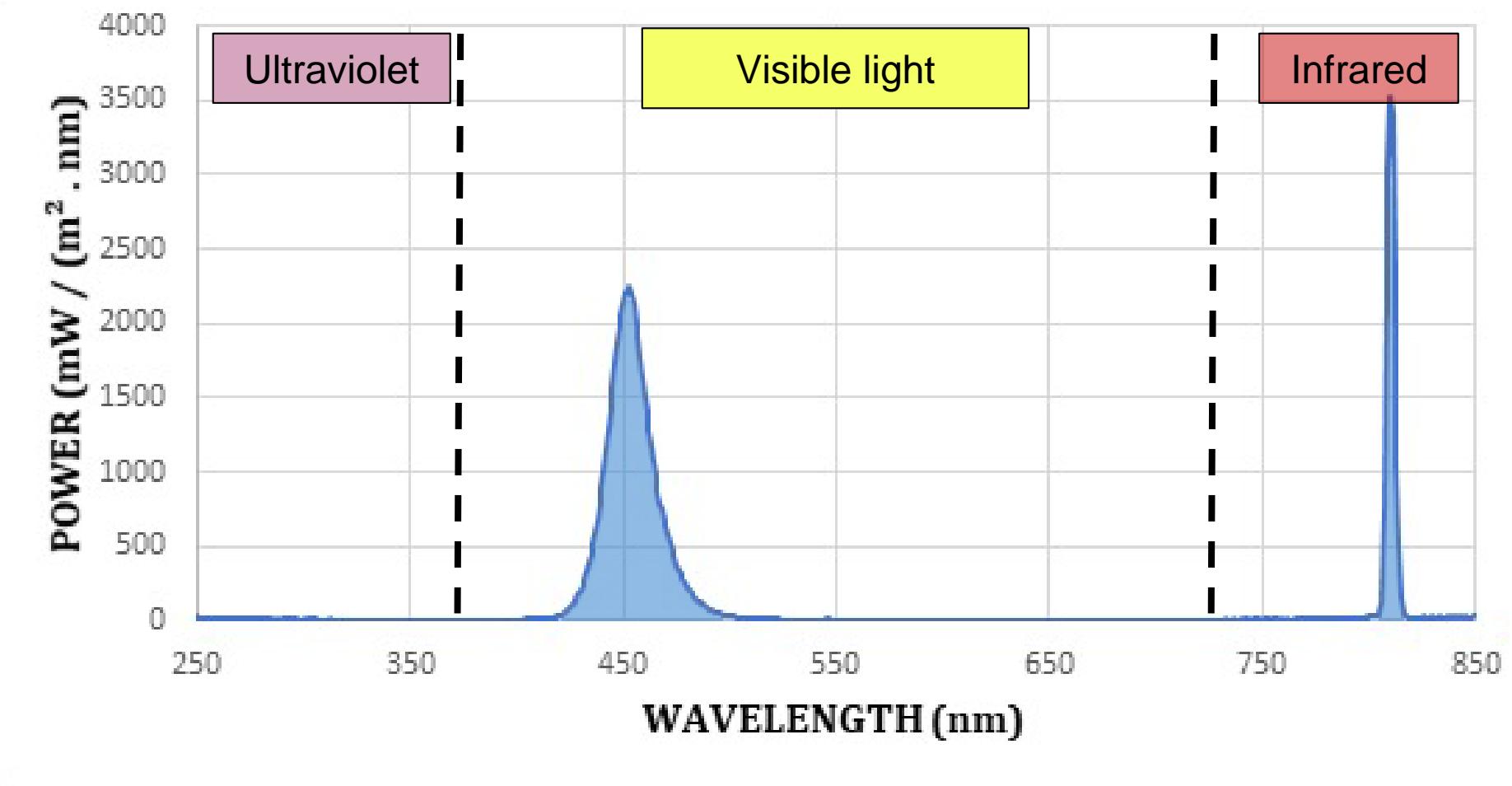
Results





<http://gg.gg/dt96v>

Results

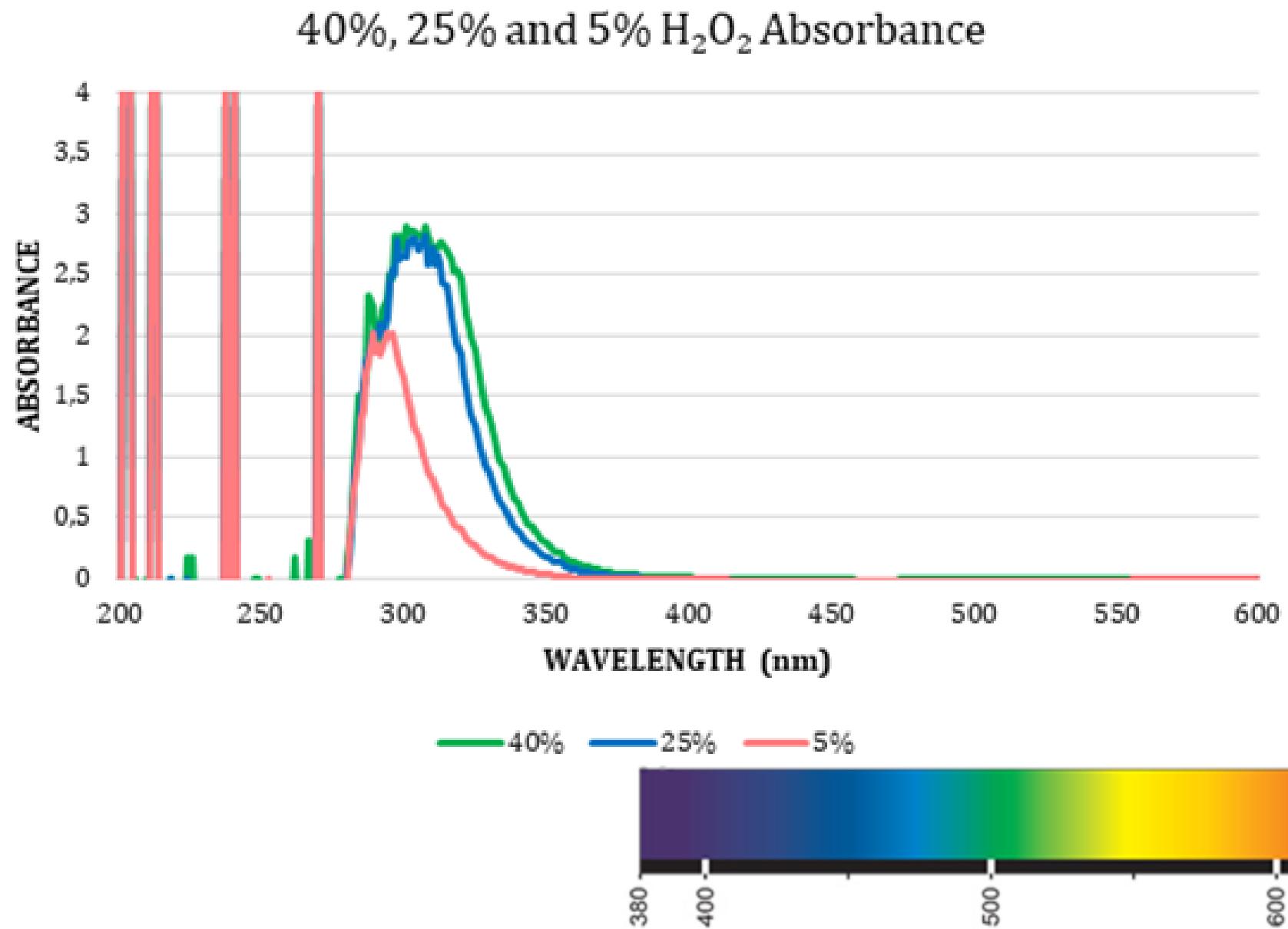


Light Spectrum	Wavelength	Power (mW/cm ²)
Visible light	$400 \text{ nm} < \lambda < 750 \text{ nm}$	5,92485
Infrared	$\lambda > 750 \text{ nm}$	1,86708
Ultraviolet	$\lambda < 400 \text{ nm}$	0,16811

3x

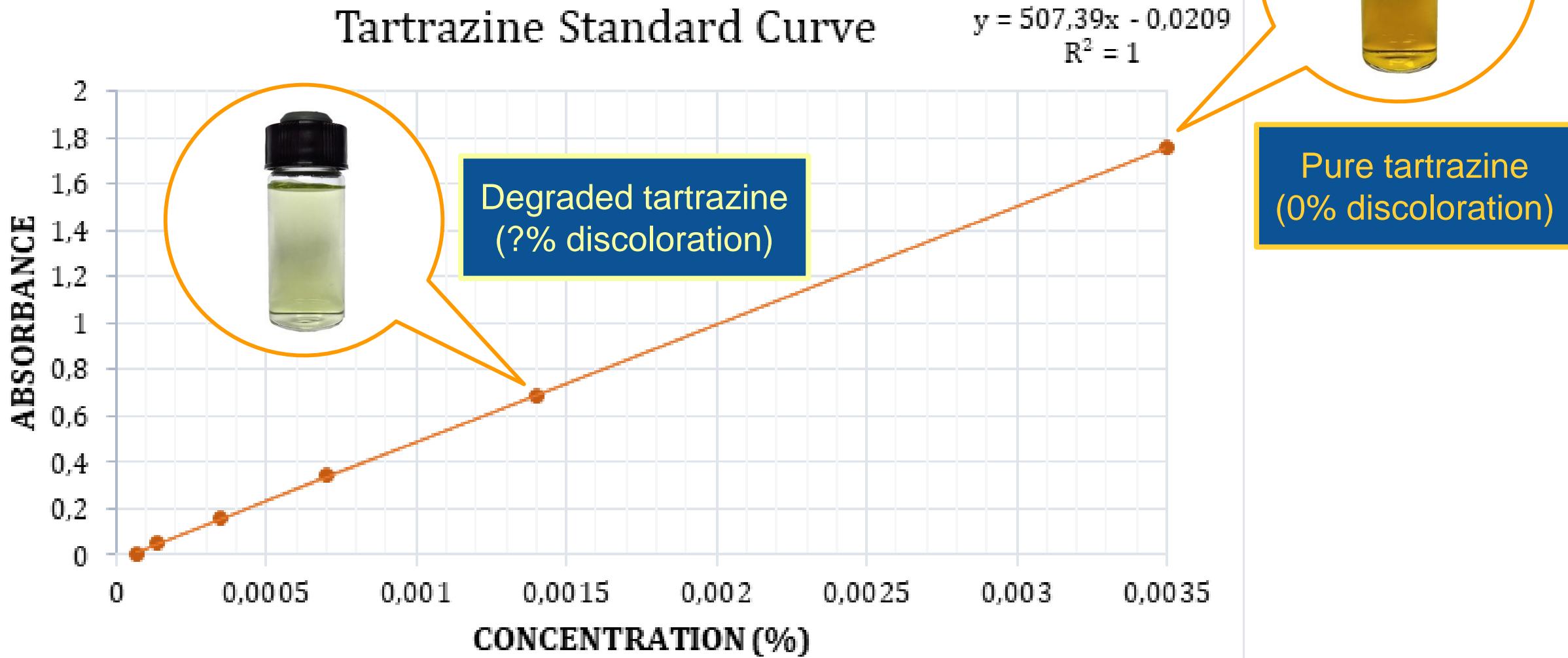
$\cong 0$

Results

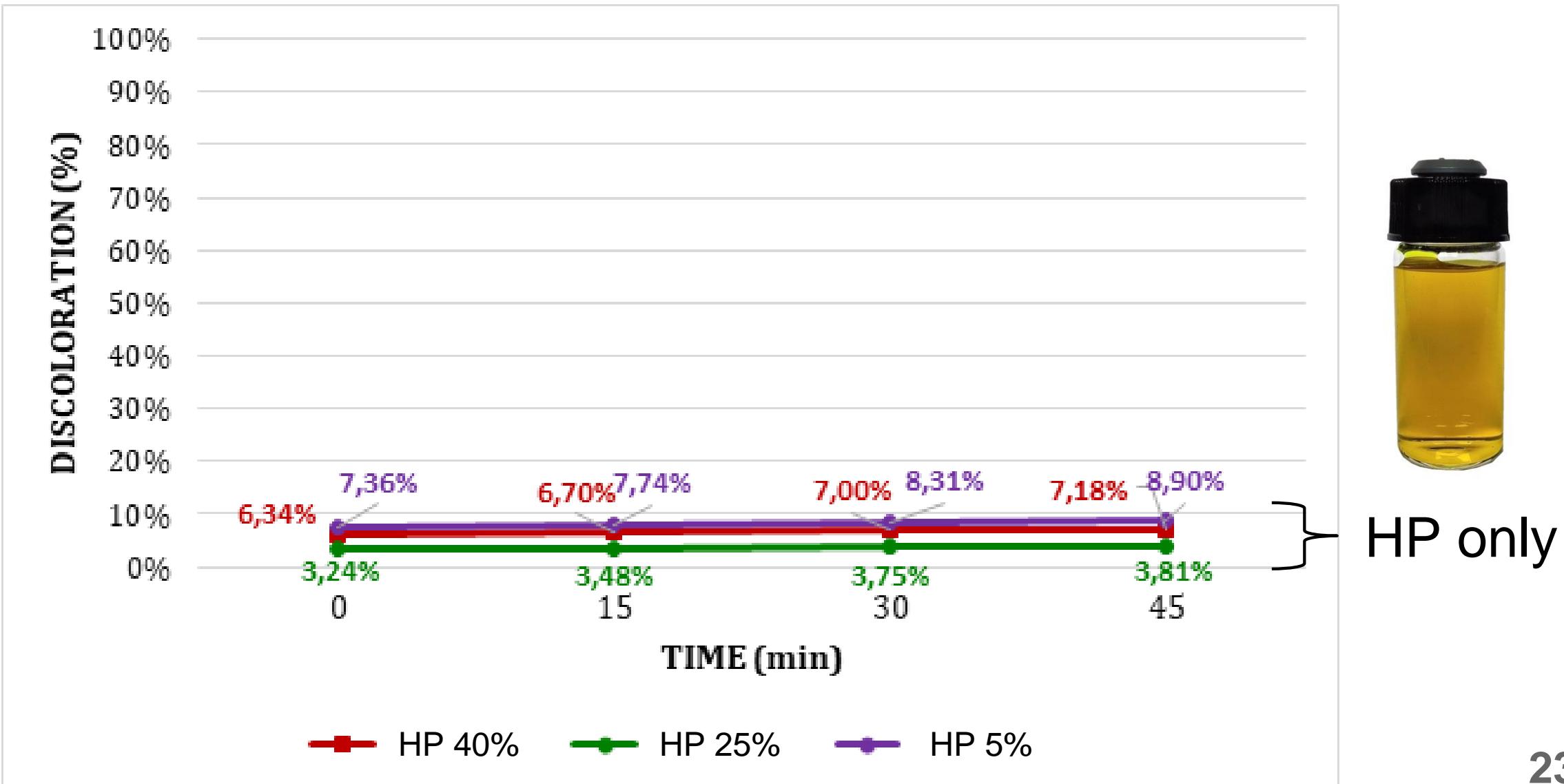


GROTTUS-DRAPER LAW

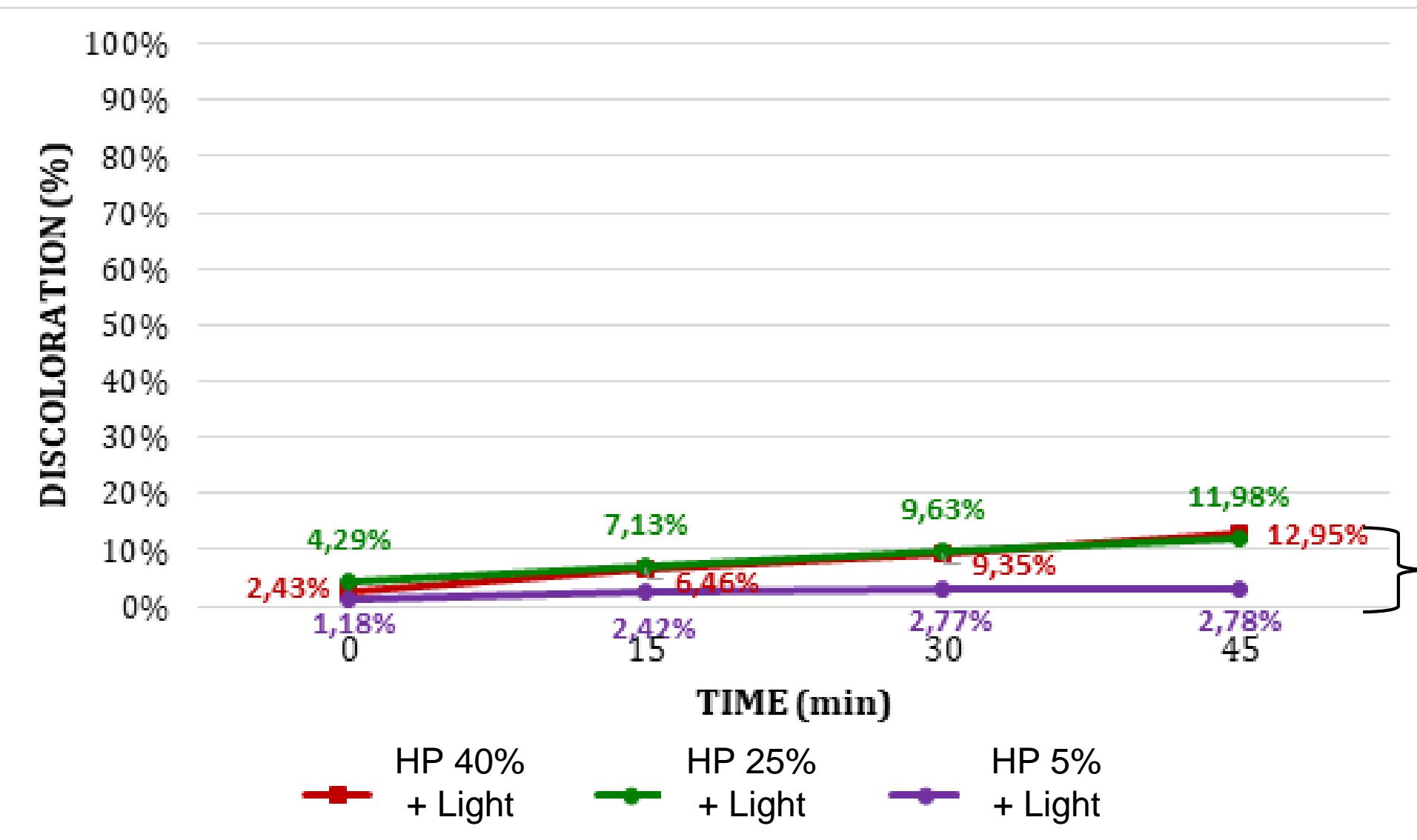
“Only that light which is **absorbed** by a system can bring about a photochemical change”



Results

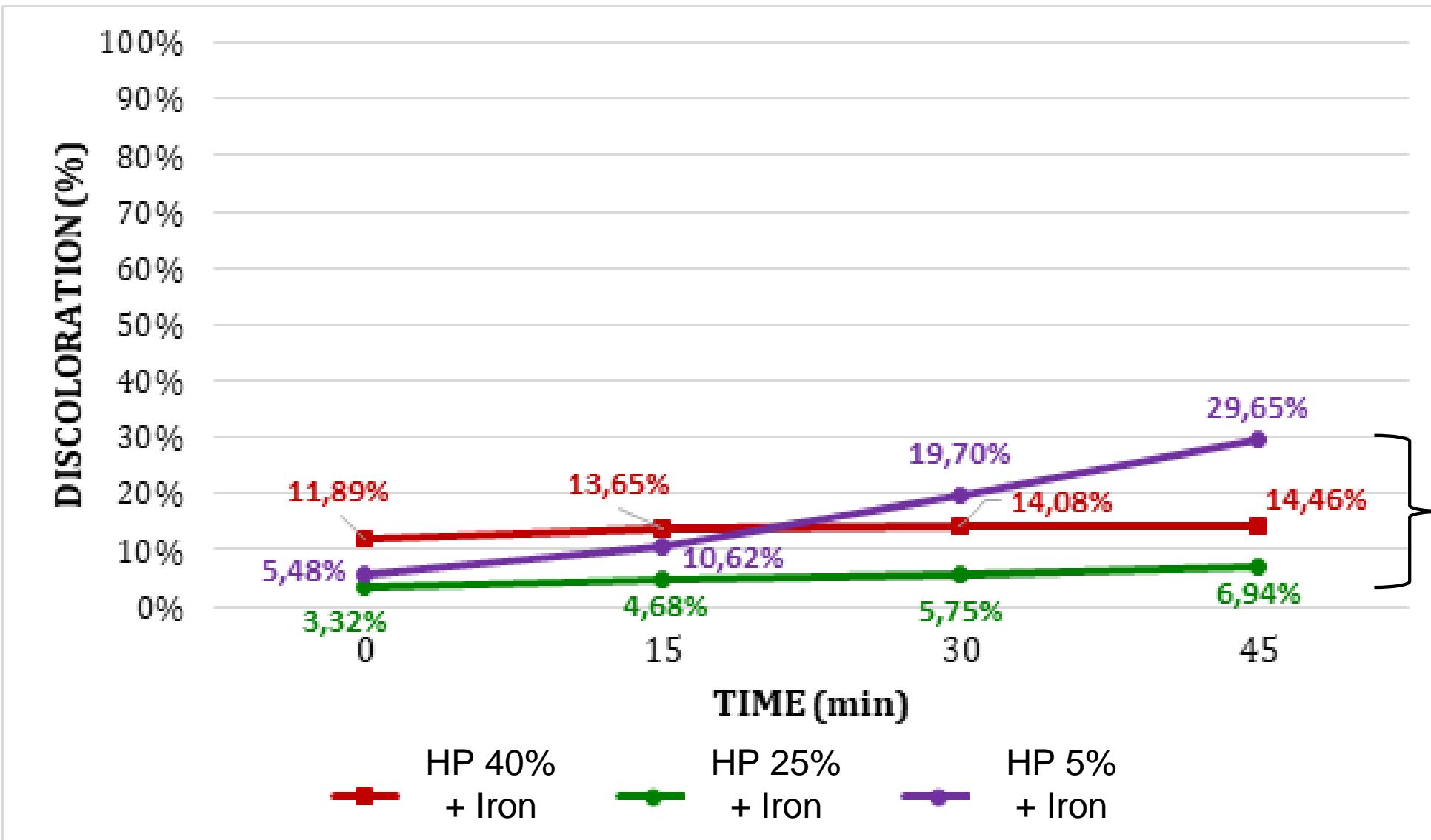


Results



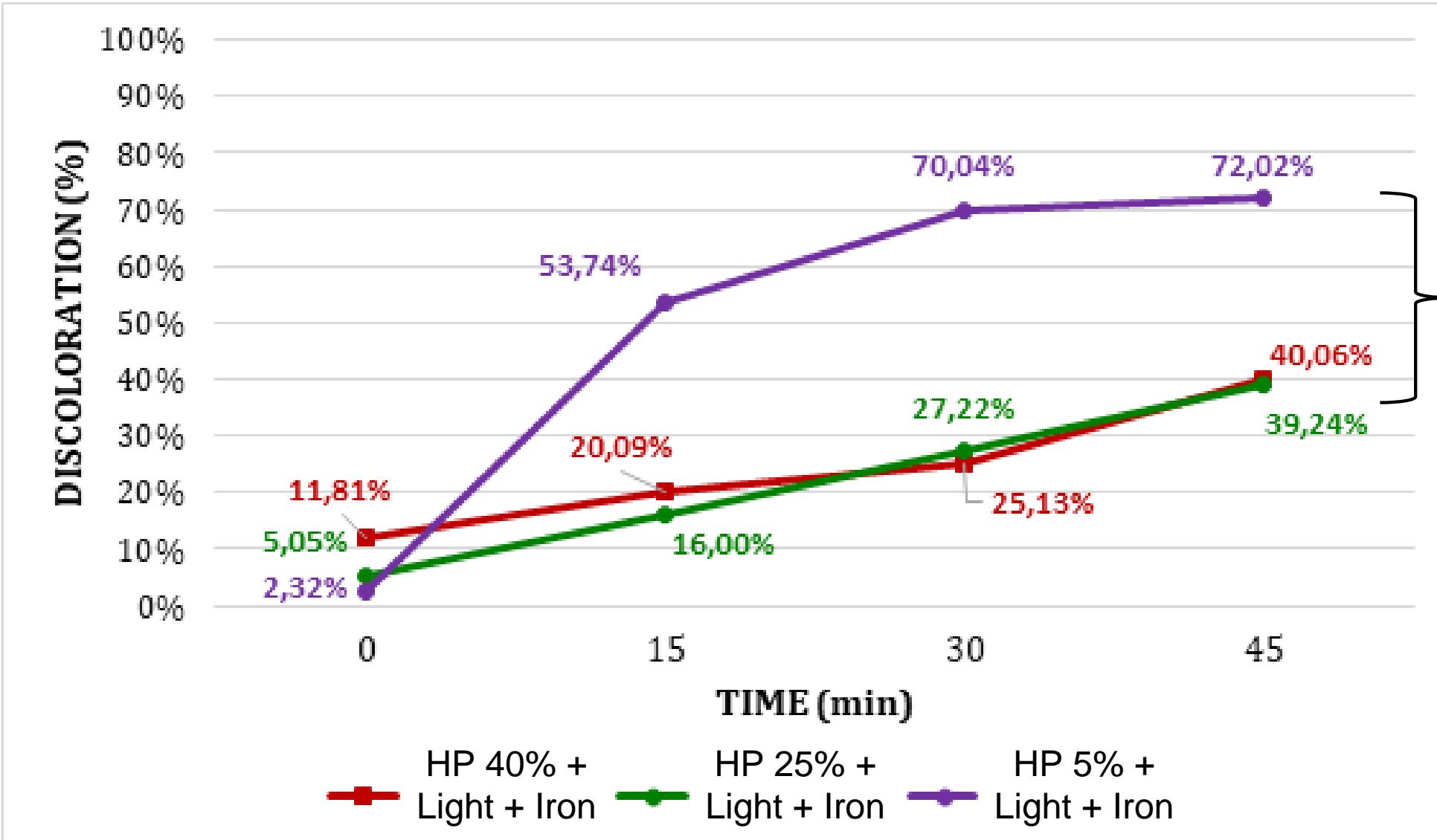
HP +
Light

Results



HP +
Iron

Results



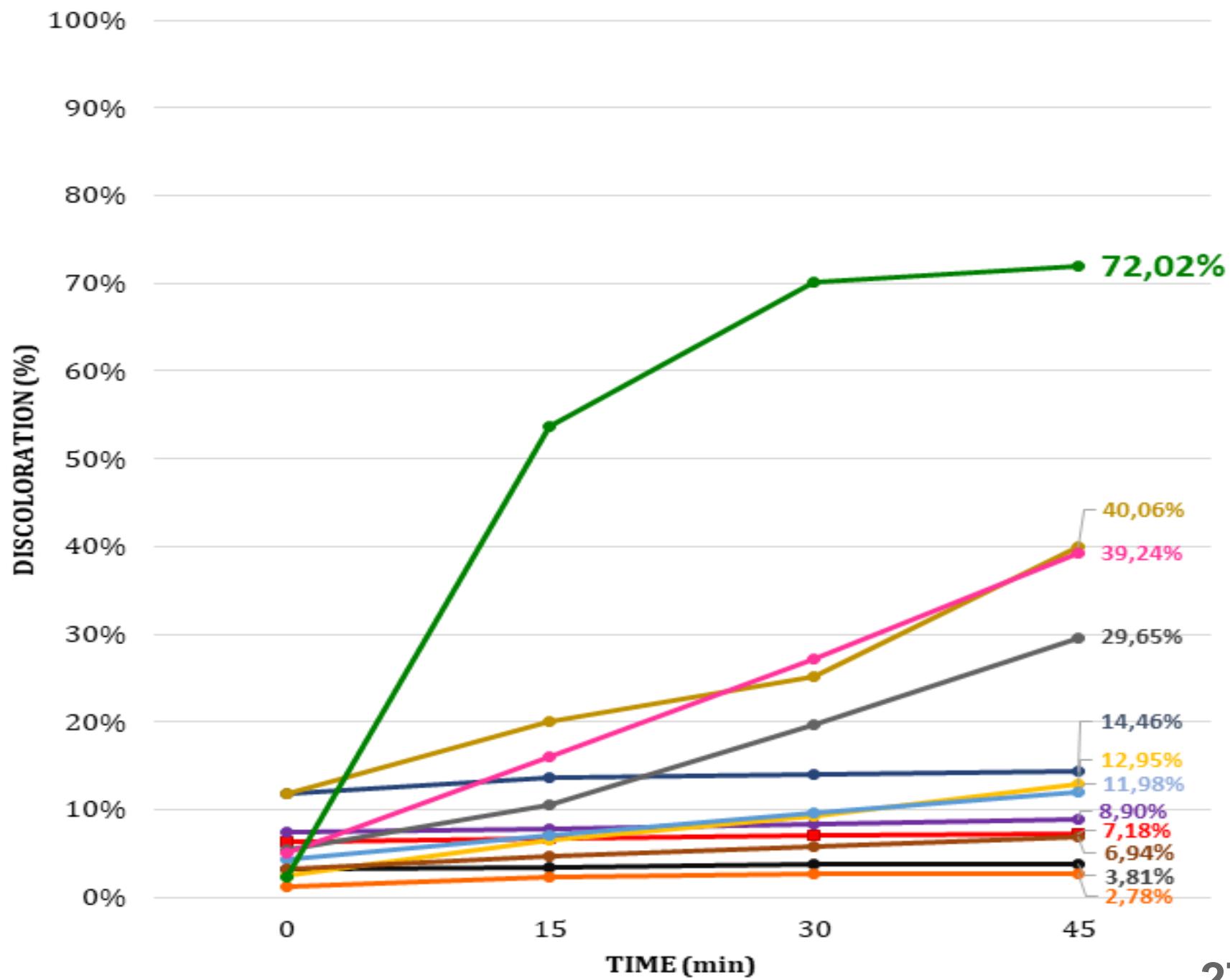
HP +
Light +
Iron

Conclusion

5% HP + Light + Iron (72% discoloration)



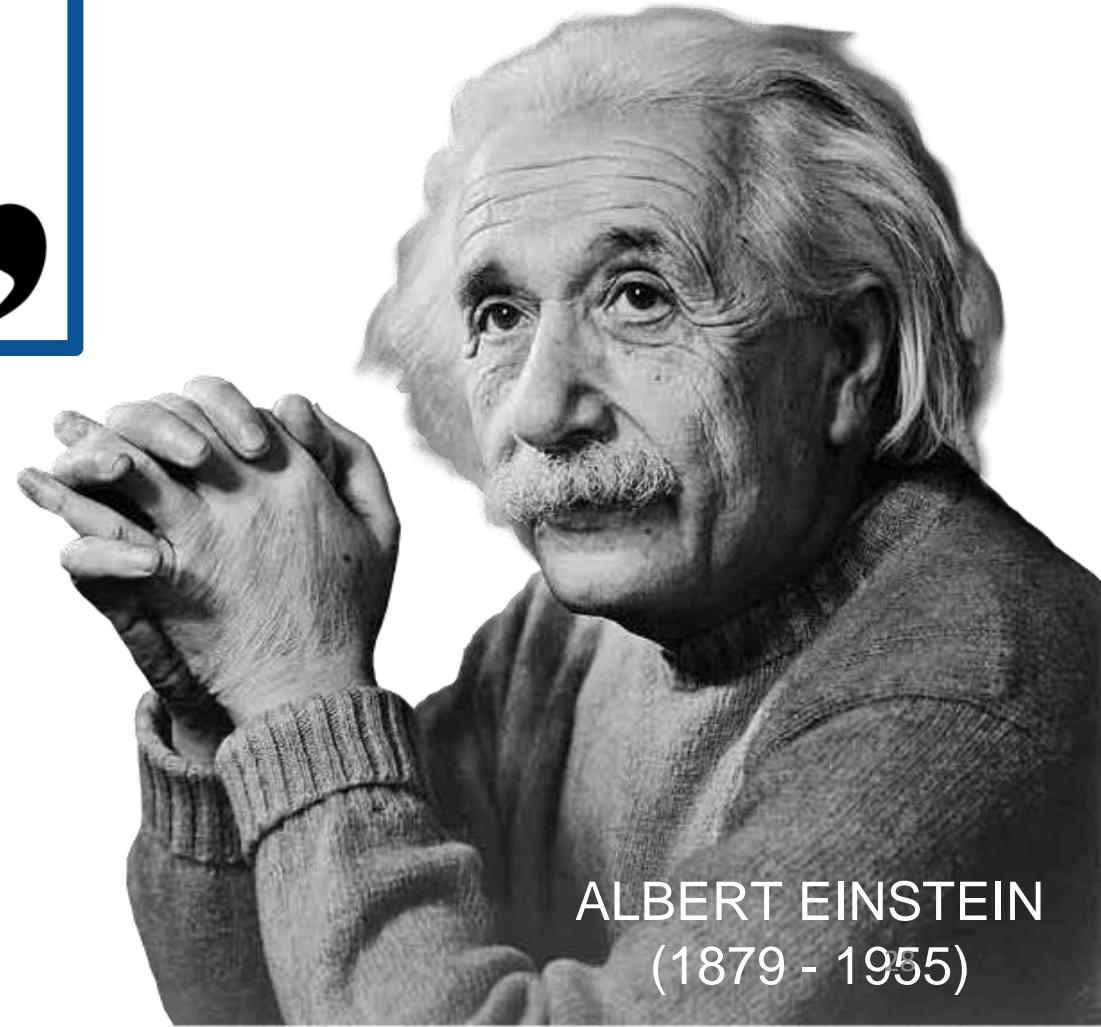
Pure tartrazine
(0% discoloration)



“

Insanity:
doing the same thing over and
over again and expecting
different results.

”



ALBERT EINSTEIN
(1879 - 1955)



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